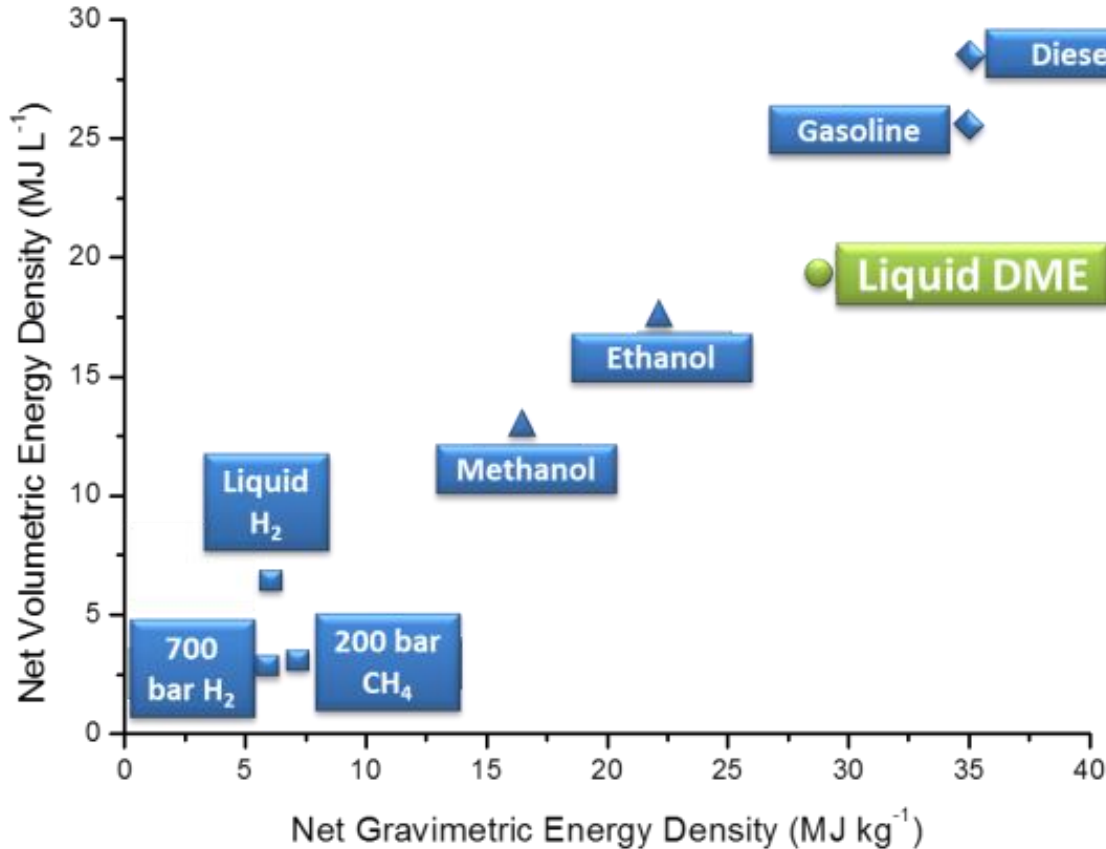


ENHANCED CONVERSION OF CO₂ FROM BIOGAS TO DIMETHYL ETHER BY IN-SITU WATER REMOVAL



Rajat Bhardwaj, Maartje Feenstra, Marco Linders, Jurriaan Boon, Juliana Monteiro, Earl Goetheer

RATIONAL: WHY BIO-DME?



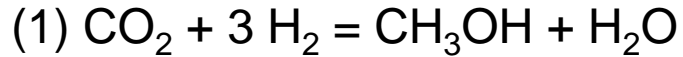
Concept: Instead of biomethane as a product stream, convert the biogas to green methanol or dimethylether.

+ Independence from gas infrastructure.

+ Larger potential for transport

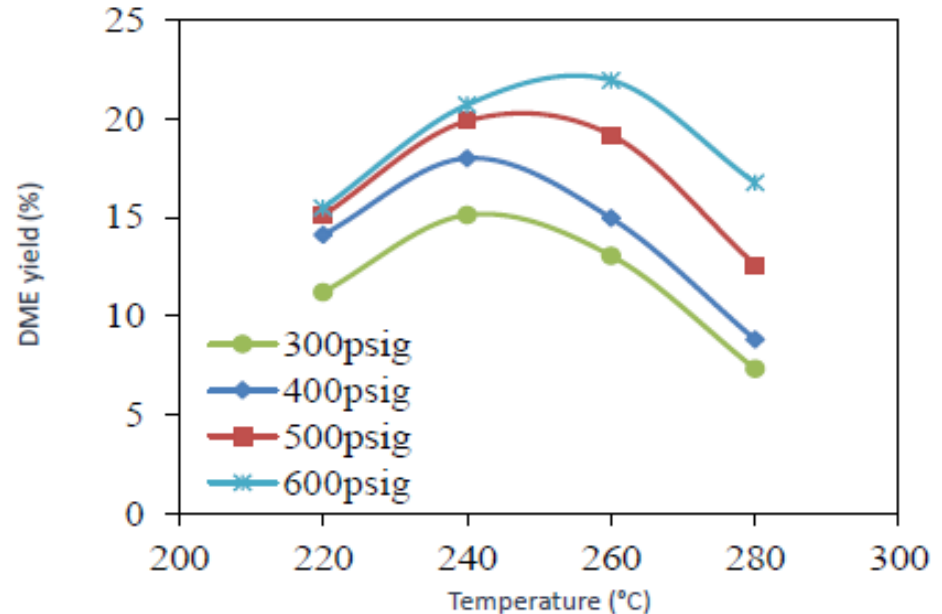
+ Higher product value

DME PRODUCTION



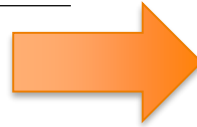
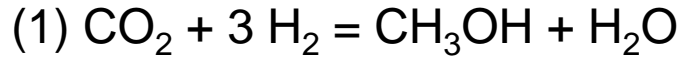
Typical conversion conditions:

T ~ 250 C, P ~ 50 bar



- › Effect of pressure
- › Trade-off between kinetics and thermodynamics

DME PRODUCTION



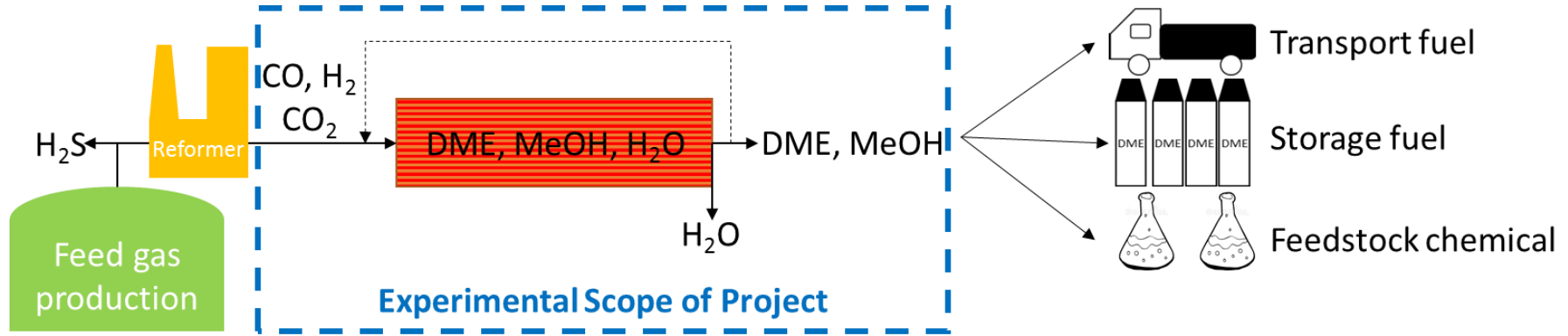
› Shift equilibrium towards DME (Le Chatelier principle):

- High pressure
- Water removal

Typical conversion conditions:

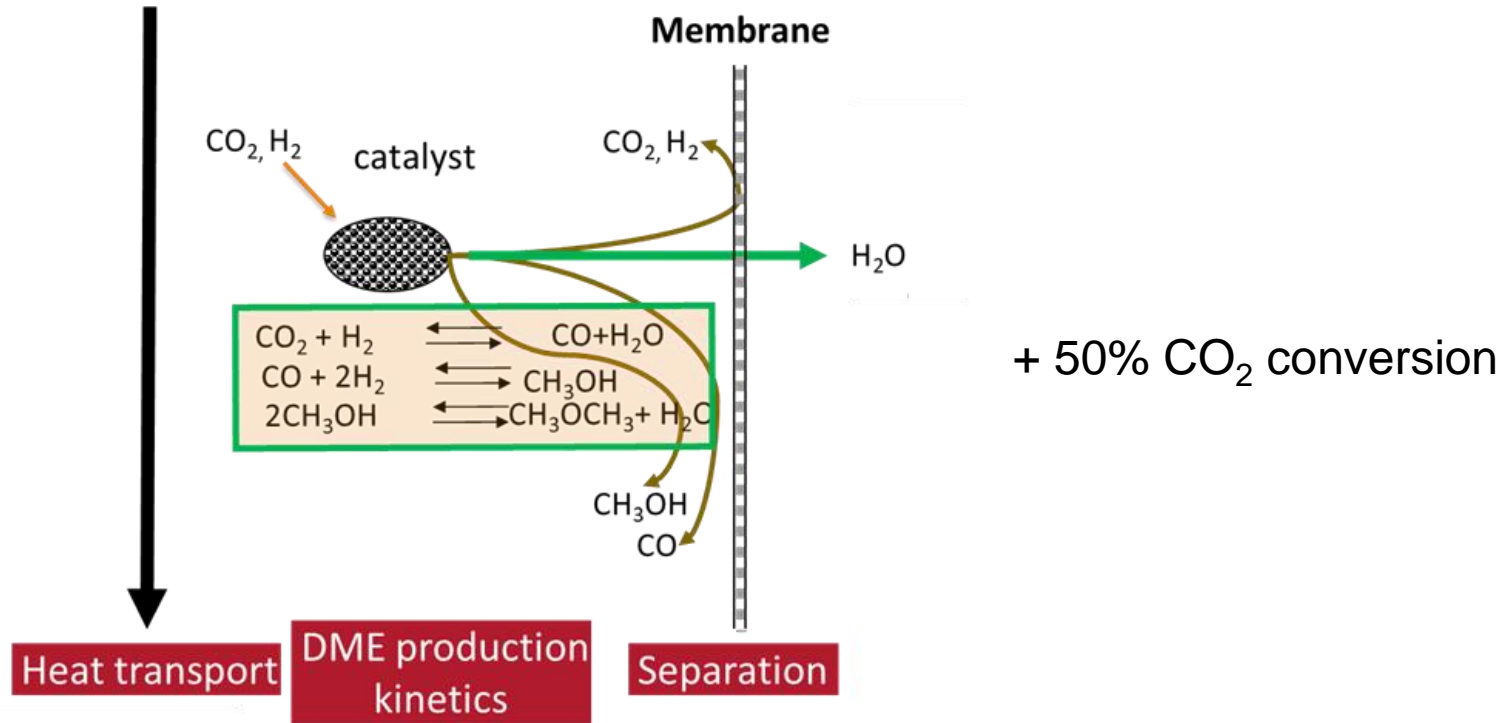
T ~ 250 C, P ~ 50 bar

IN-SITU CONVERSION FOR ENHANCED DME PRODUCTION

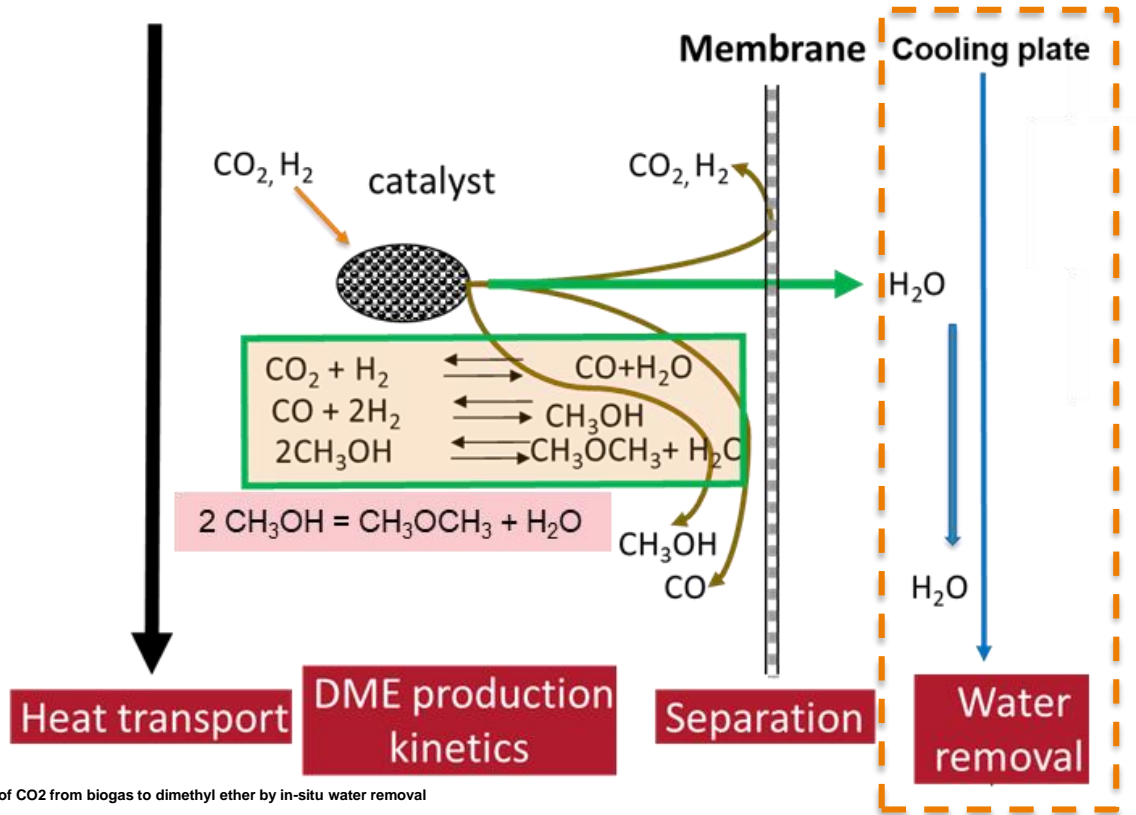


New process concept to produce DME from CO_2 rich gasses such as biogas

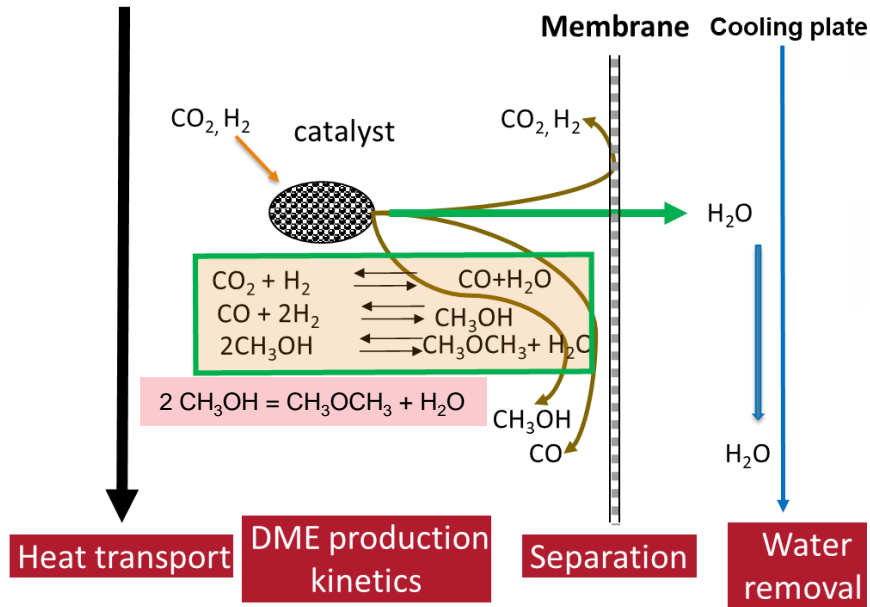
ON-GOING WORK ON METHANOL PRODUCTION



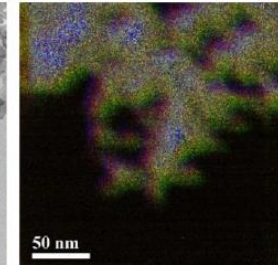
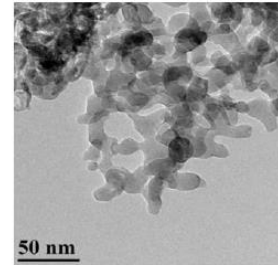
REACTOR CONCEPT



REACTOR CONCEPT



Existing bi-functional catalysts for DME production



Colors represent:
 Cu - yellow
 Zn - blue
 Al - green
 O - red

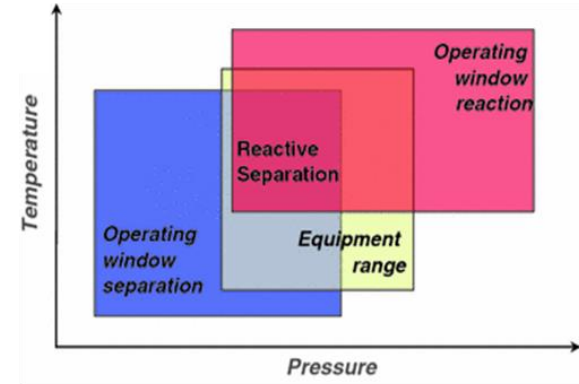
Catalyst Properties

- 🔧 Methanol synthesis: CuO/ZnO/Al₂O₃
- 🔧 DME synthesis: H-ZSM-5
- 🔧 BET surface area 132 m²/g
- 🔧 Particle size 10-30 nm

Ref: GTI's ARPA-E REFUEL Project

REACTOR CONCEPT

- › Choice of system conditions (P, T) is dependent on a several competing factors



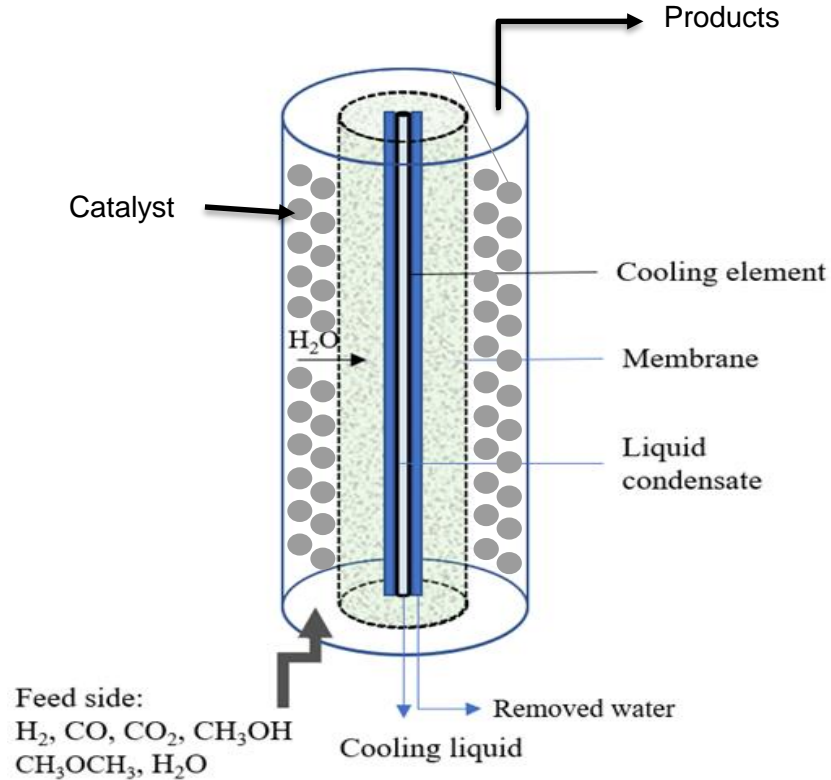
System design: T,P

Membrane selection

Transport (Heat and Mass)

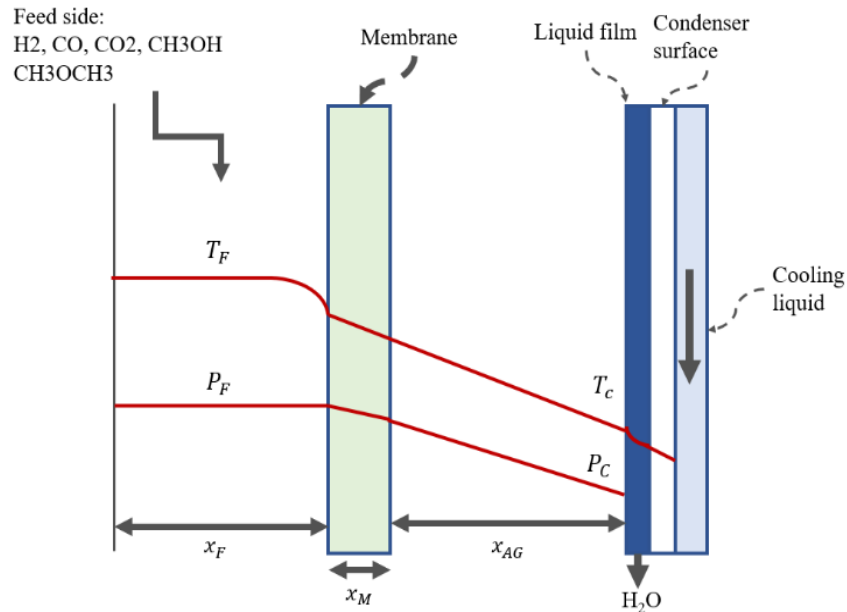
Reaction kinetics

REACTOR CONCEPT



REACTOR MODEL DESCRIPTION

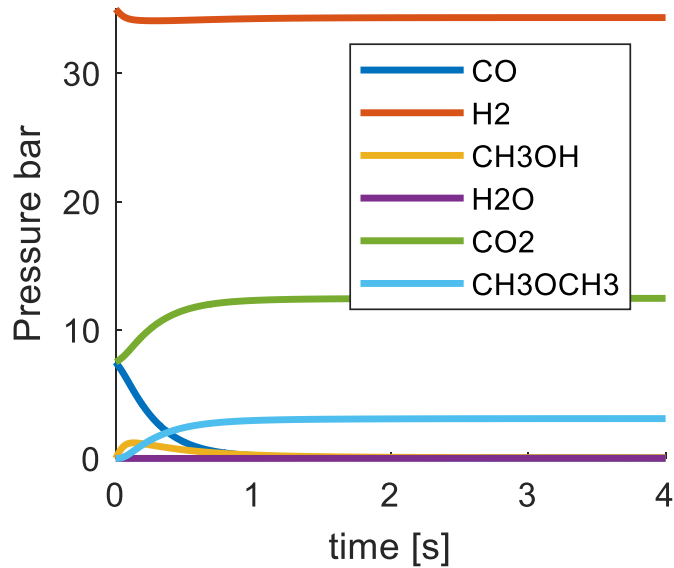
Mass transfer model



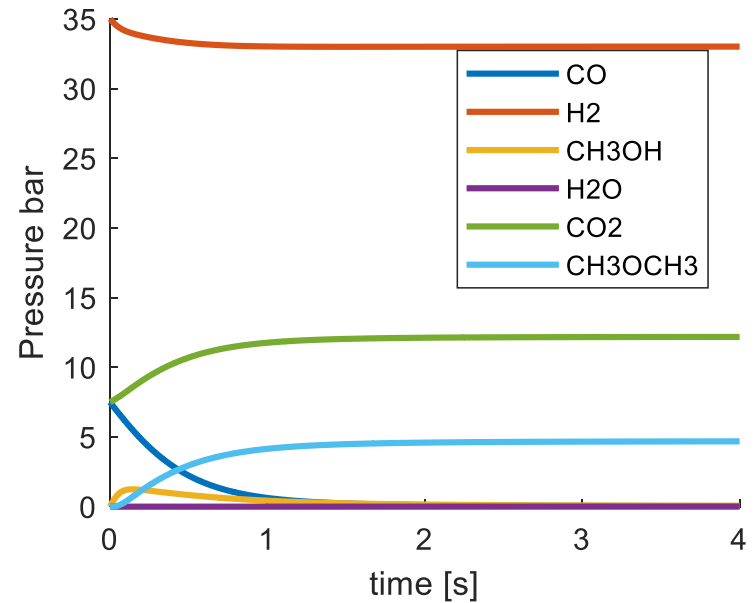
Mass transfer model with reactor kinetics for estimation of DME production

MODELLING RESULTS

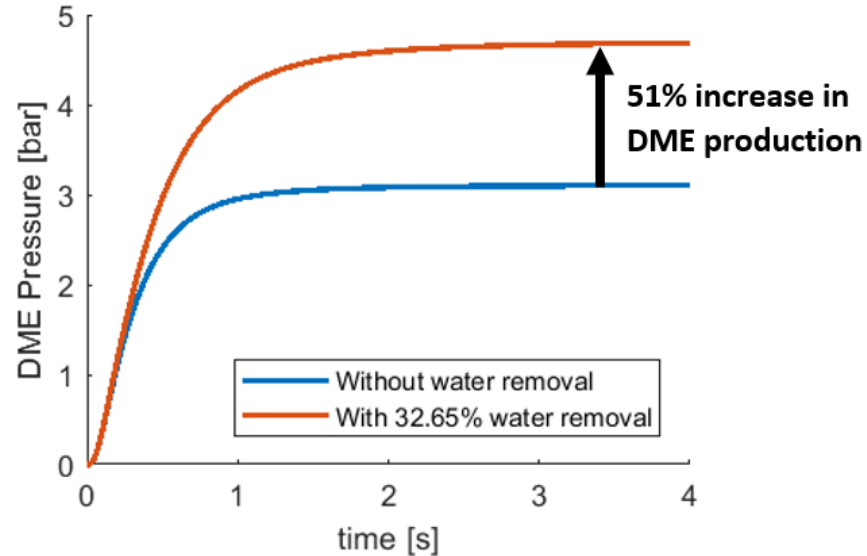
without water removal



with Water removal



MODELLING RESULTS



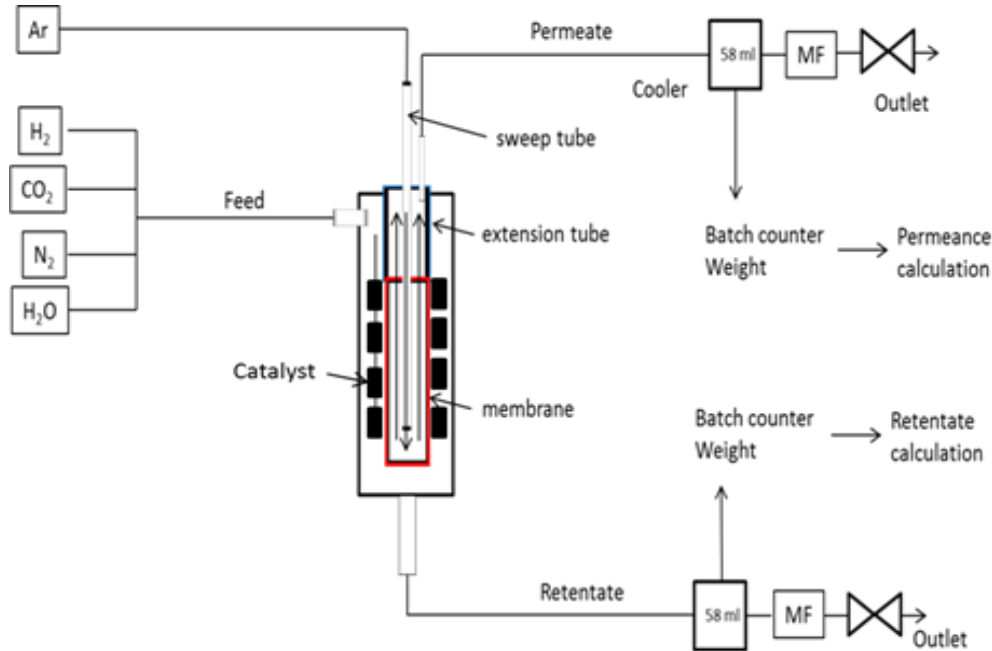


› EXPERIMENTAL WORK

EXPERIMENTAL SETUP



PERFORMANCE MEASUREMENT SETUP



Conditions:

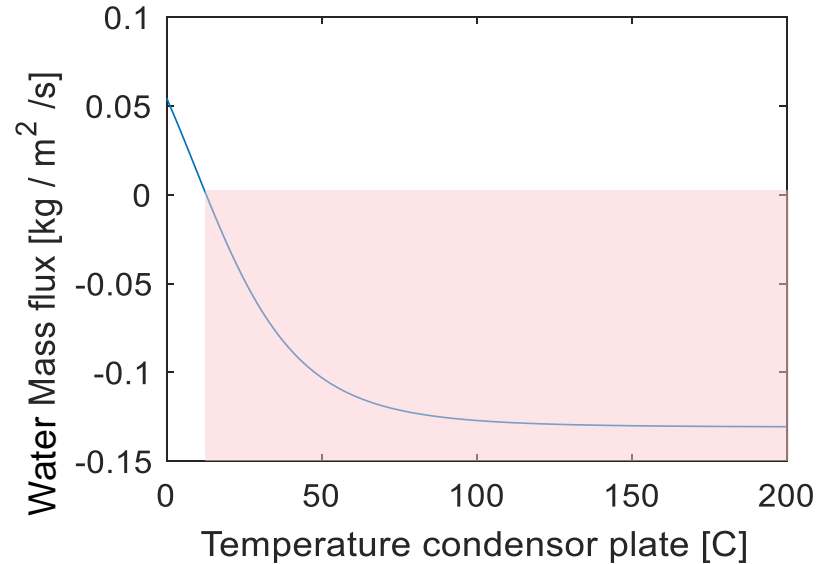
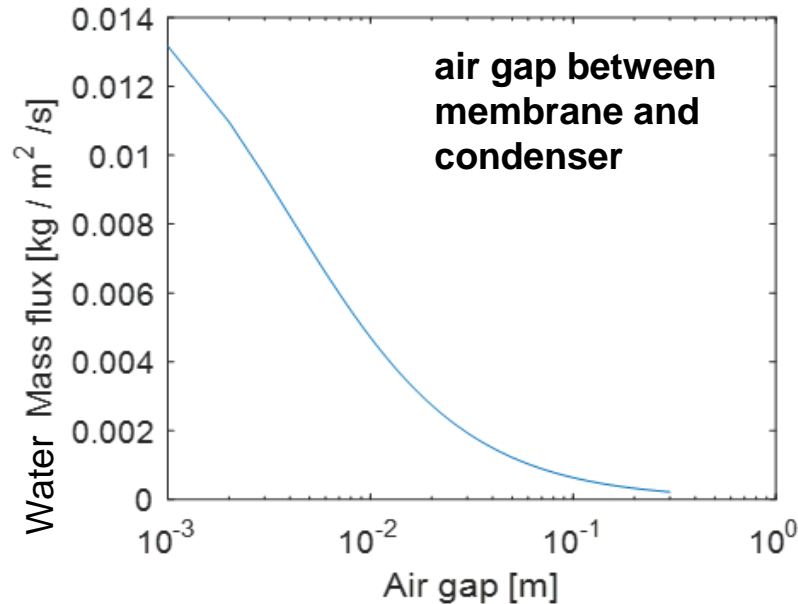
$T = 220\text{ }^{\circ}\text{C}$; $P = 25\text{ bar}$

Flux results:

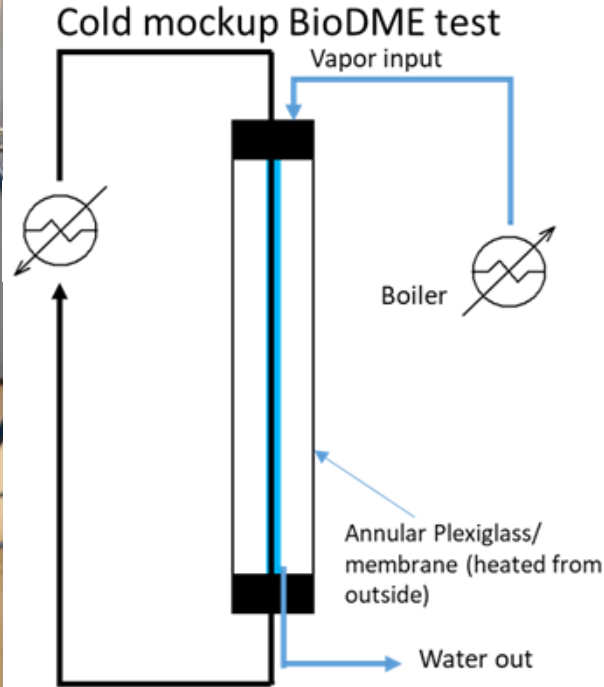
- $1\text{-}2 \cdot 10^{-5}\text{ mol/m}^2 \cdot \text{h} \cdot \text{Pa}$ for H₂
- $2 \cdot 10^{-4}\text{ mol/m}^2 \cdot \text{h} \cdot \text{Pa}$ for H₂O

Gas	Quantity in L _n /min	Max g/h
Hydrogen (H ₂)	20	107
Nitrogen (N ₂)	20	1500
Carbon dioxide (CO ₂)	10	1179
Carbon monoxide (CO)	2	150
Water (H ₂ O)	3	144
Methanol (CH ₃ OH)	1.7	144
Sweep Ar or He	20 (or 15)	2140 (or 160)

OPTIMIZATION FOR WATER REMOVAL BASED ON CRITICAL PARAMETERS



PRACTICAL TESTS: MEMBRANE/ CONDENSER FIT



And thin film movement along the majority of length of condenser

CONCLUSIONS

- › Model estimates an increase of up to 51% in the production of DME
- › Condenser Temperature and air gap are critical parameters for system design.
- › Membrane characterization show selectivity of H₂O to H₂ of up to 16
- › In-situ condenser configuration practically tested for dimensions of reactor.

A nighttime photograph of a city street. In the foreground, a modern, curved pedestrian bridge or walkway with a glass railing is illuminated. The background shows multi-story buildings with lit windows and green light trails from a moving vehicle or light source. The overall scene is urban and contemporary.

› **THANK YOU FOR YOUR
ATTENTION**

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ECN ›

TNO

innovation
for life

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