

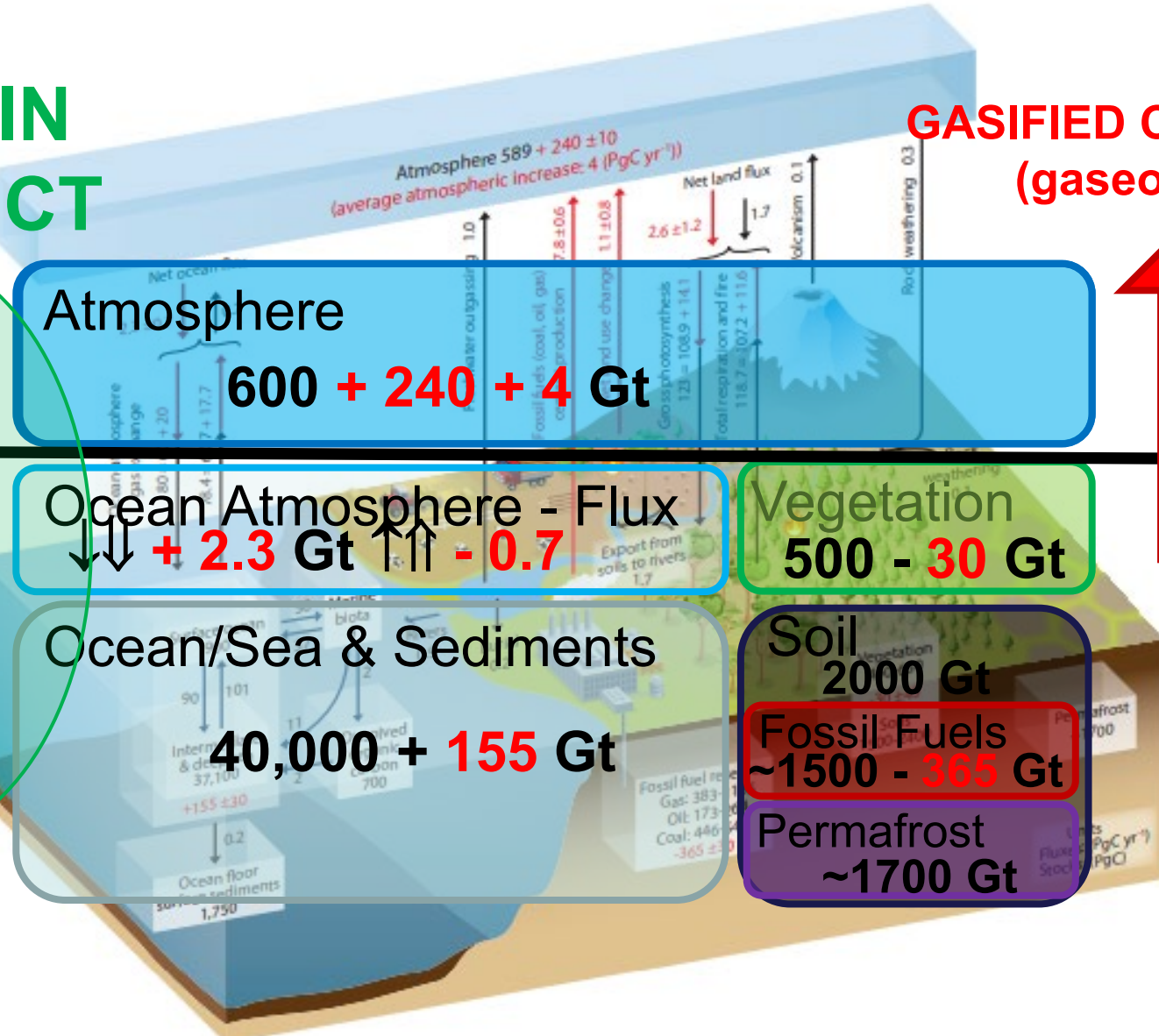
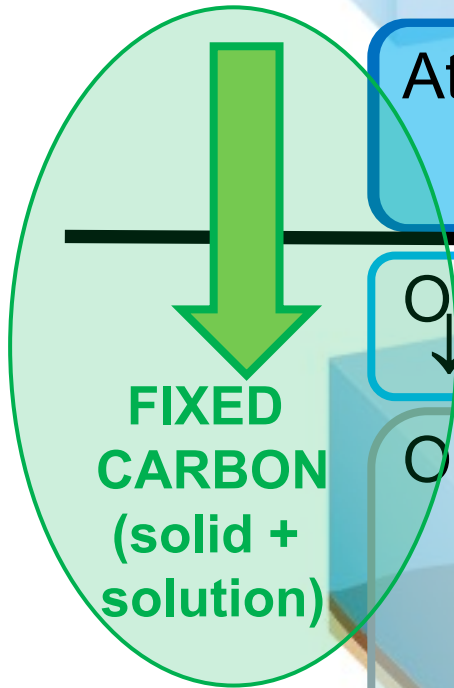
FUNMIN

FUNdamental Studies of MINeral Carbonation with Application to CO₂ Utilisation



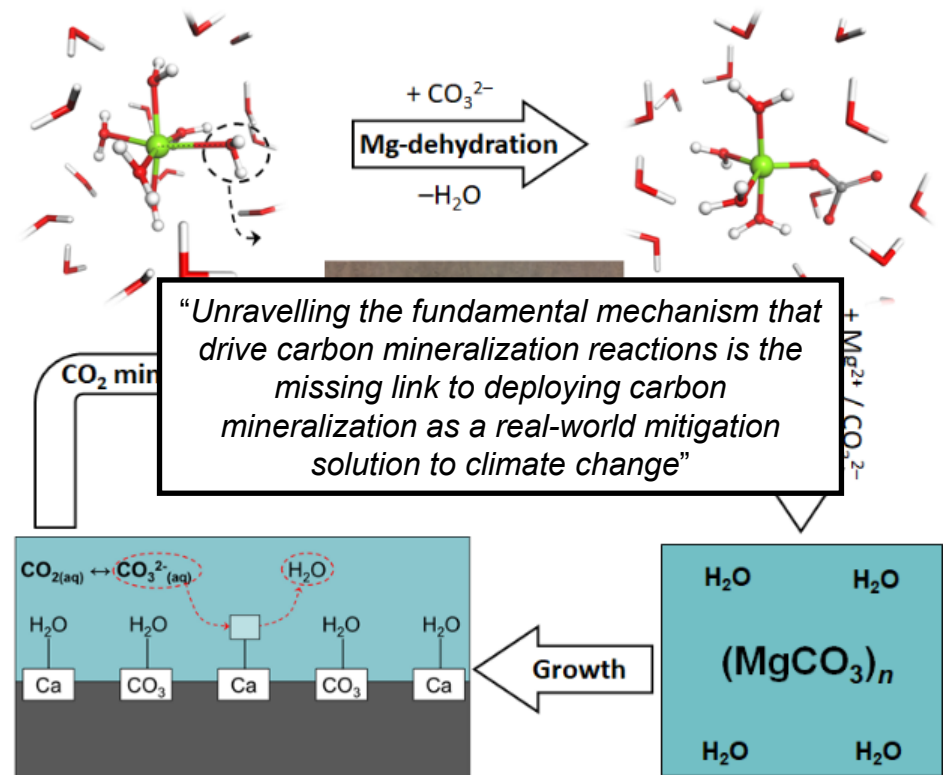
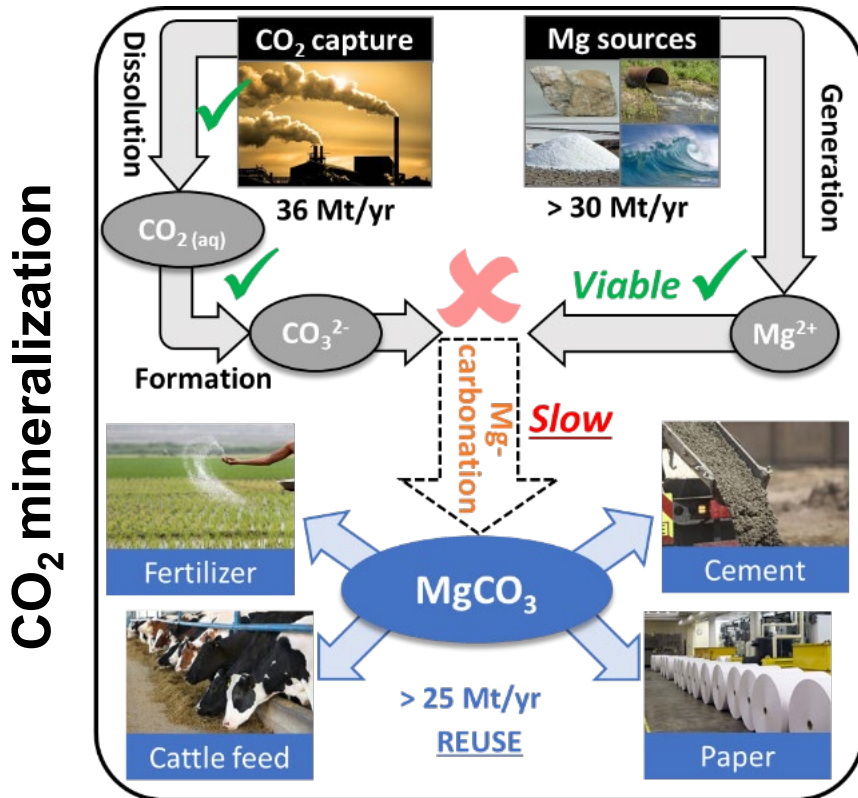
Carbon Reservoirs & Fluxes + Human Fluxes

FUNMIN PROJECT



CO₂ into solid AS added-value products

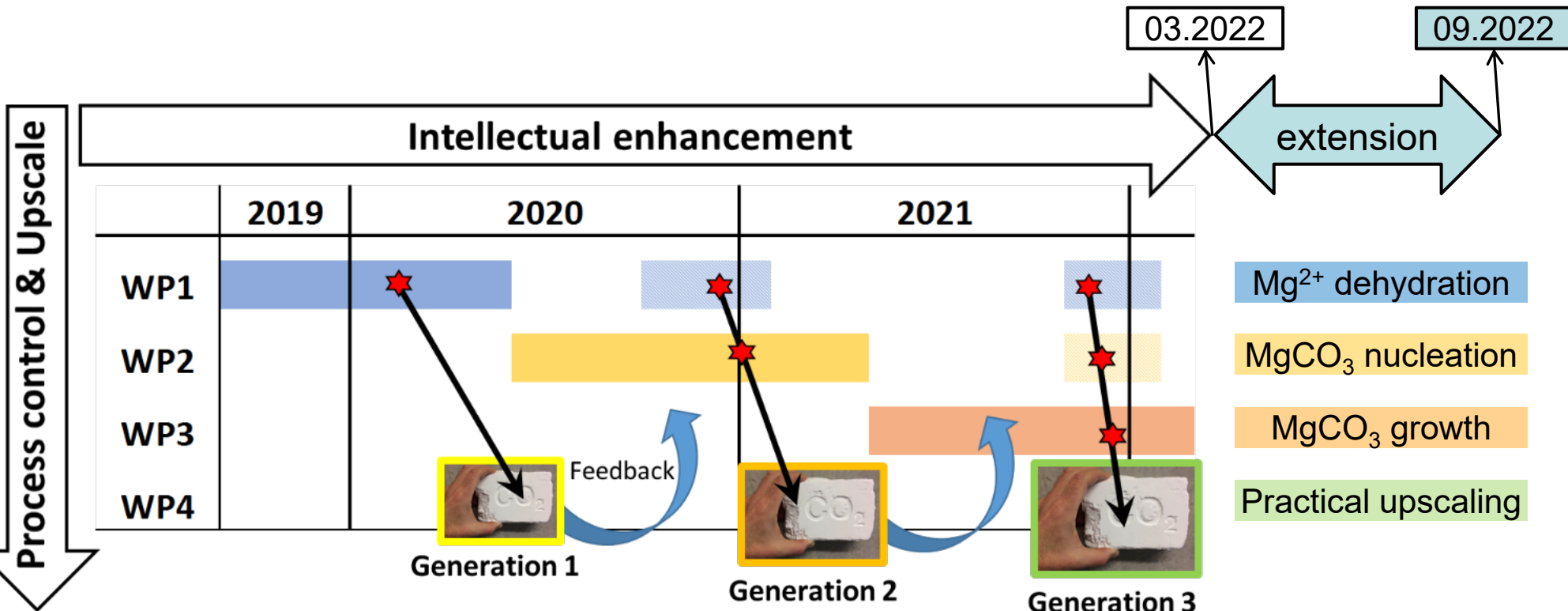
“CCUS can create new industries and markets using carbon dioxide, such as chemicals, plastics, and building materials” *



* The UK carbon capture, usage and storage (CCUS) deployment pathway, BEIS, 2018

Project implementation

Core activities (**WP1-3**) to characterize the **molecular processes** controlling MgCO_3 crystallization; applied component (**WP4**) to **optimise** conditions

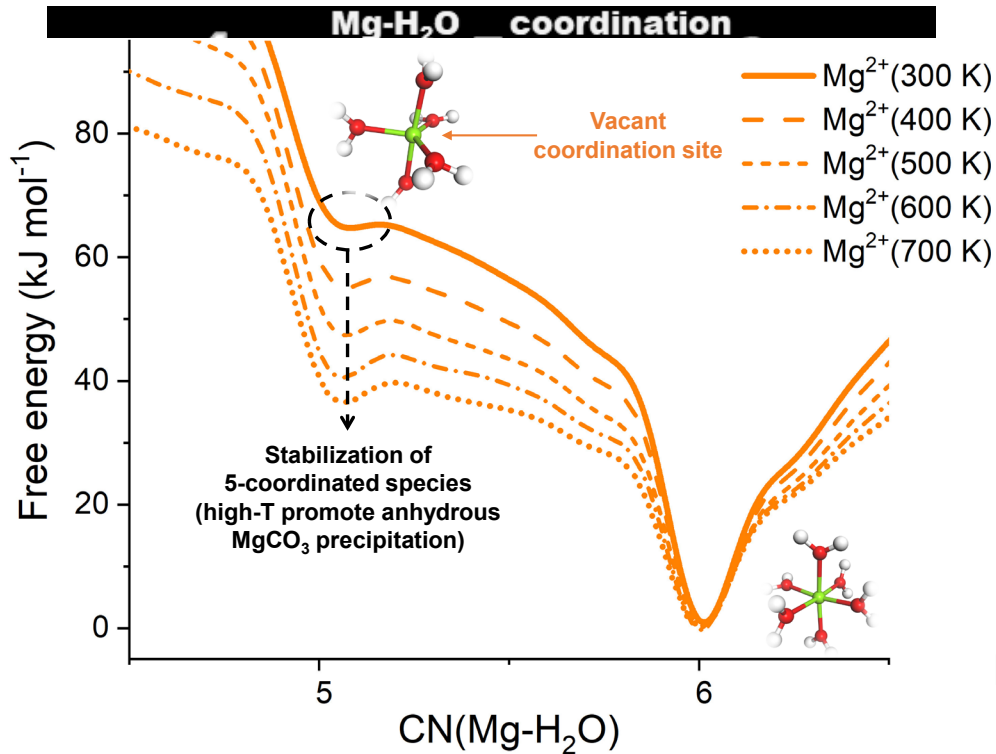


Interaction between the scientific (**WP1-3**) and practical (**WP4**) components of the project.
WP1: Mg-dehydration; **WP2**: MgCO₃ nucleation; **WP3**: Magnesite Growth; **WP4**: Upscaling

Key achievements

1. Control rate-determining Mg^{2+} dehydration
2. Database of additives promoting MgCO_3 formation
3. Empirical tracking of additives' effect on MgCO_3 production
4. Mechanism of MgCO_3 production & form
5. (amorphous-to-crystalline transition)
6. Flow cell(s) for “real time” watching CO_2 mineralization
7. Pre-industrial bulk reactor for generating 10s of kg of carbonates

WP1: Processes controlling Mg^{2+} dehydration



CrystEngComm HOT article:
“New insights into the role of solution additive anions in Mg^{2+} dehydration: implications for mineral carbonation”

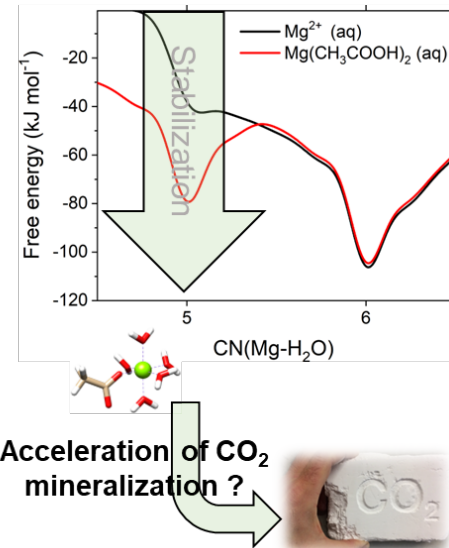
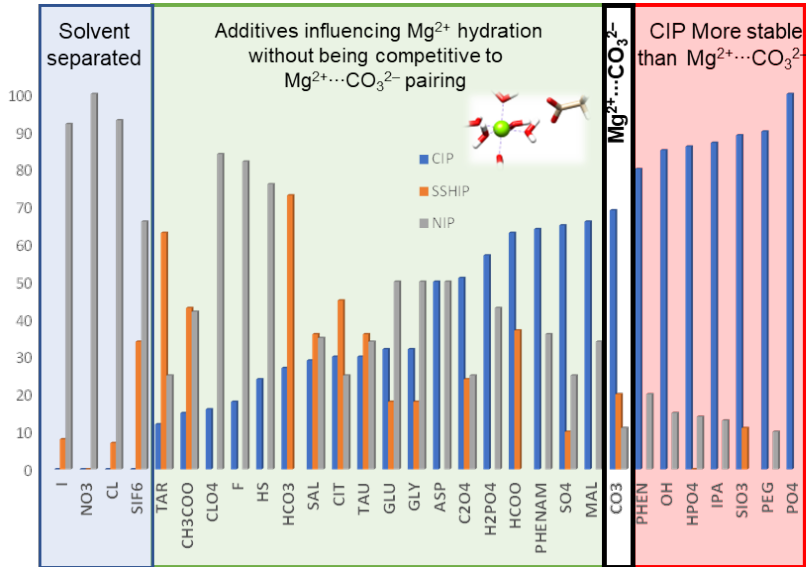
Key points:

- $Mg(H_2O)_6^{2+}$ only stable coordination in pure water
- Additives help initiate mineralisation
- Promotion of low-temperature mineralisation

Front cover

CrystEngComm, 2021, 23, 4896-4900

WP1: Computational database of solution additives



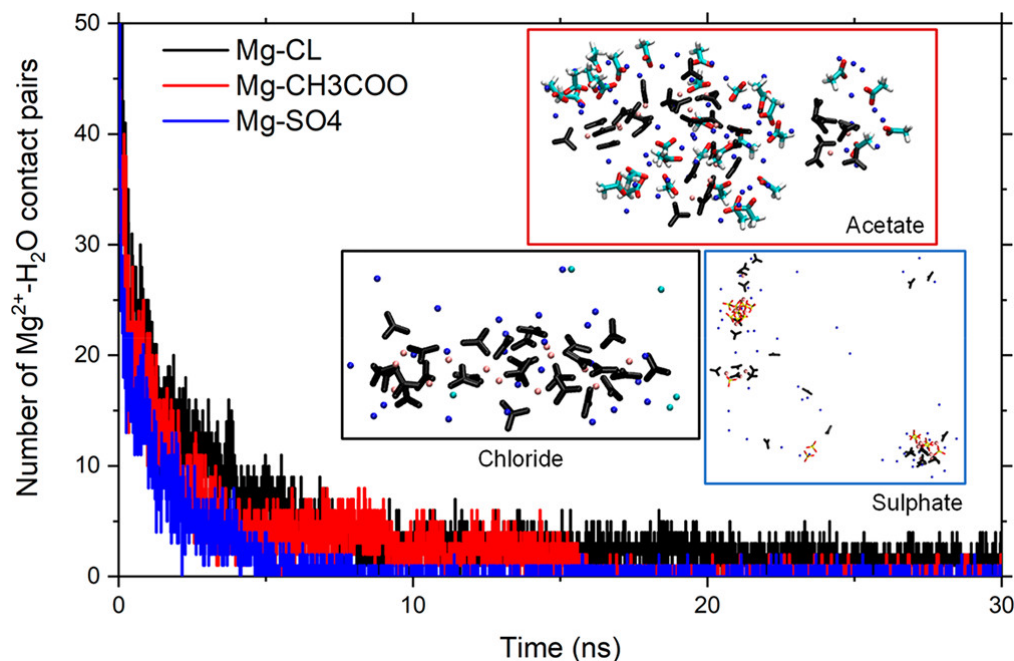
**Non-competitive ions
promoting
Mg²⁺ dehydration**

Acceleration of CO₂ mineralization ?

Key points:

- Extensive computer simulations of the early stages of MgCO₃ nucleation in the presence of solution additives
- Characterised the ability of thirty additives to promote dehydration based on well-defined molecular level criteria
- Identification of solution composition conditions catalysing the low-temperature CO₂ conversion into MgCO₃ → **promotion of CO₂-mineralisation at 25° C**

WP2: Effect of solution additives on MgCO_3 aggregation

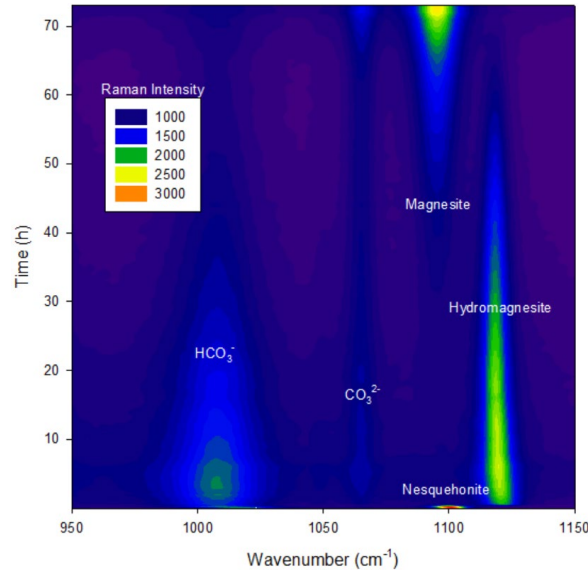
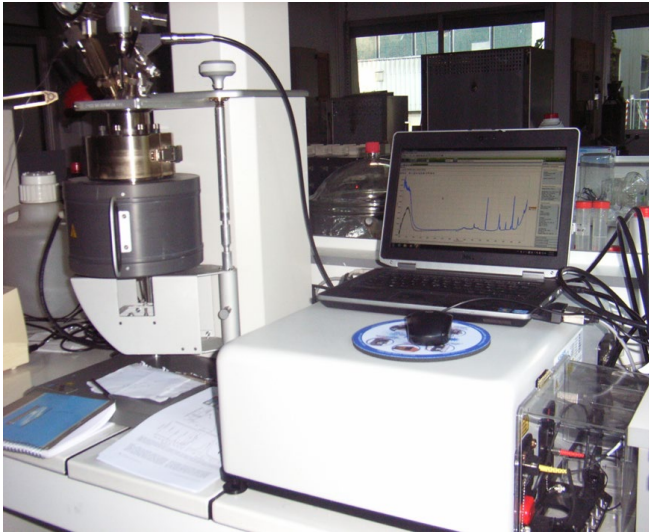


**Non-competitive ions
promoting MgCO_3 crystal
nucleation**

Key points:

- **Formation of stable carbonates with non-hazardous additives** (acetate, chloride, and sulphate)
- **Agreement between simulation ↔ experiment**

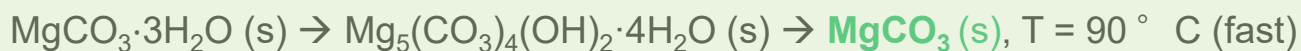
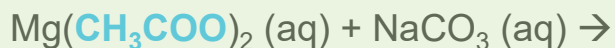
WP2: Experimental verification of MgCO_3 growth



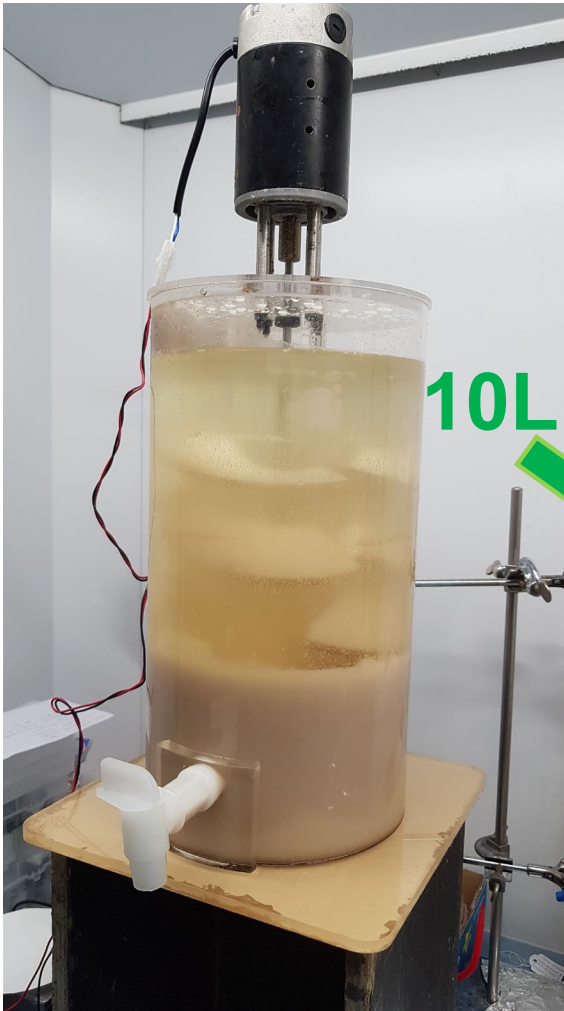
Key points:

- Real-time nucleation measurements under different solution conditions (CH_3COO^- , SO_4^{2-} , NO_3^- , ...)
- Classical nucleation pathway revealed
- The presence of acetate ions promotes “low” T MgCO_3 formation

Raman monitoring during nucleation and growth of MgCO_3 particles and crystals under hydrothermal conditions



WP4: Scale-up of CO₂-mineralisation pre-industrial scale & control of Mg-carbonate growth



- 1st generation bulk CO₂-mineralisation reactor built . Has 10L capacity, generates carbonates @ ~1-2kg / hour
- Modulating conditions and solution chemistry generates differing forms → allowing for control of particle sizing and properties

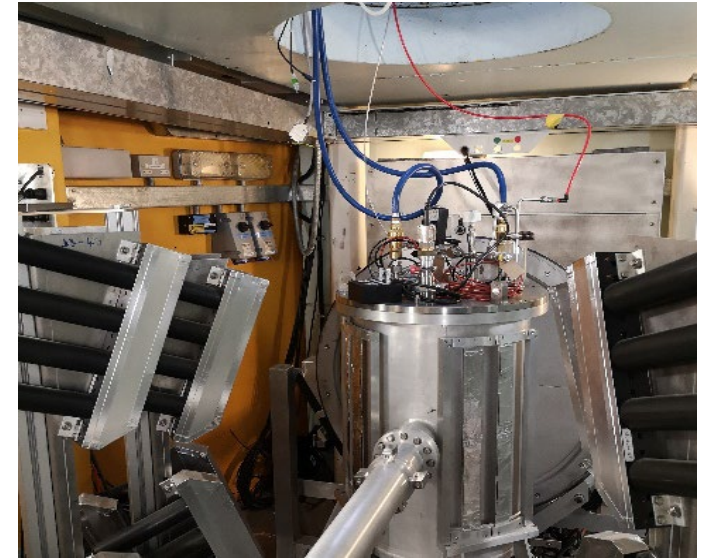
1-2kg / hr



WP4: Tracking CO₂ mineralisation from nano- to bulk-scales – Neutron beam analyses in real-time



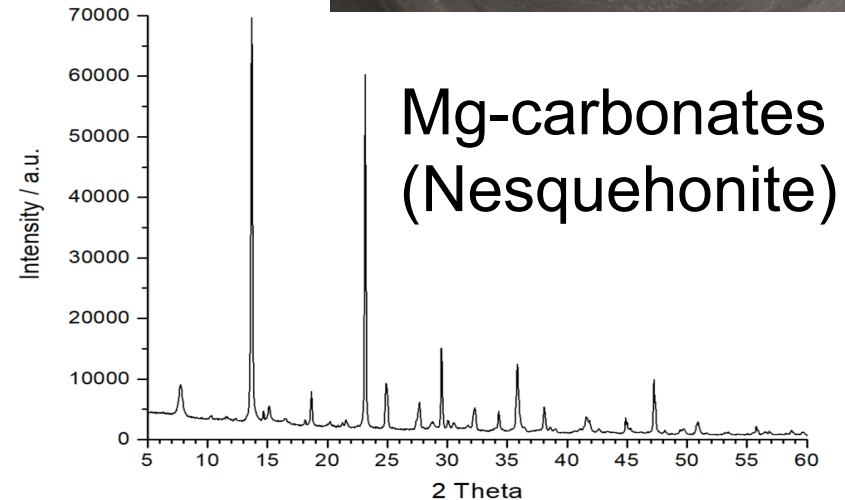
May-July 2021, RAL, UK



WP4: Mg-Carbonates produced @ ~50-100g scales



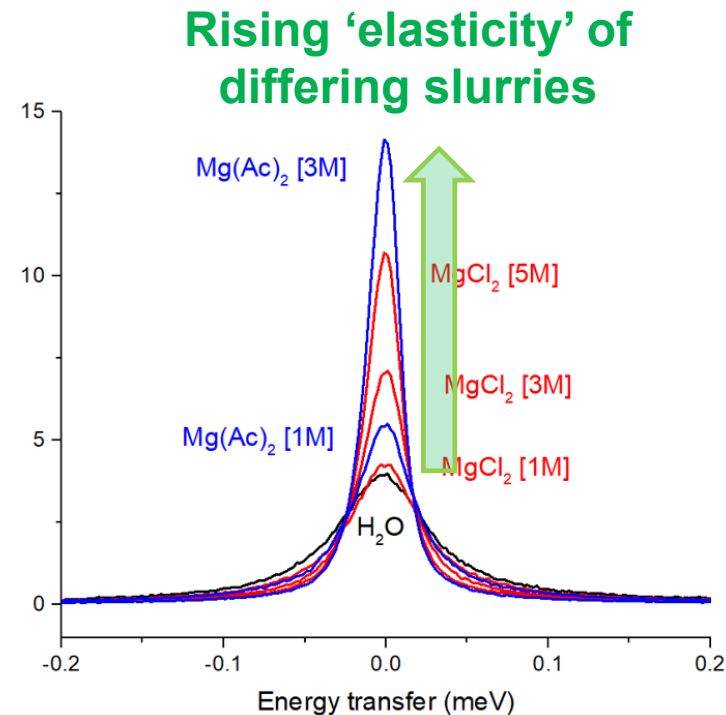
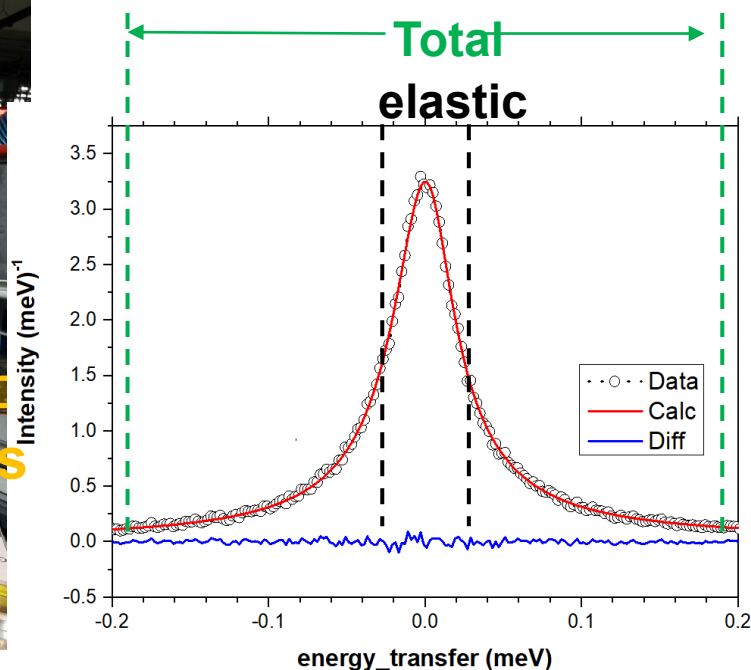
Residual
Carbonates



Key results:

- Mineralisation reactor retained industrial conditions and product homogeneity, residual carbonates in reactor likewise easily processed (removed) confirming reactor re-use (i.e., industrial robustness)
- Output product Mg-carbonates (i.e., nesquehonite) was confirmed as pure via XRD measurements

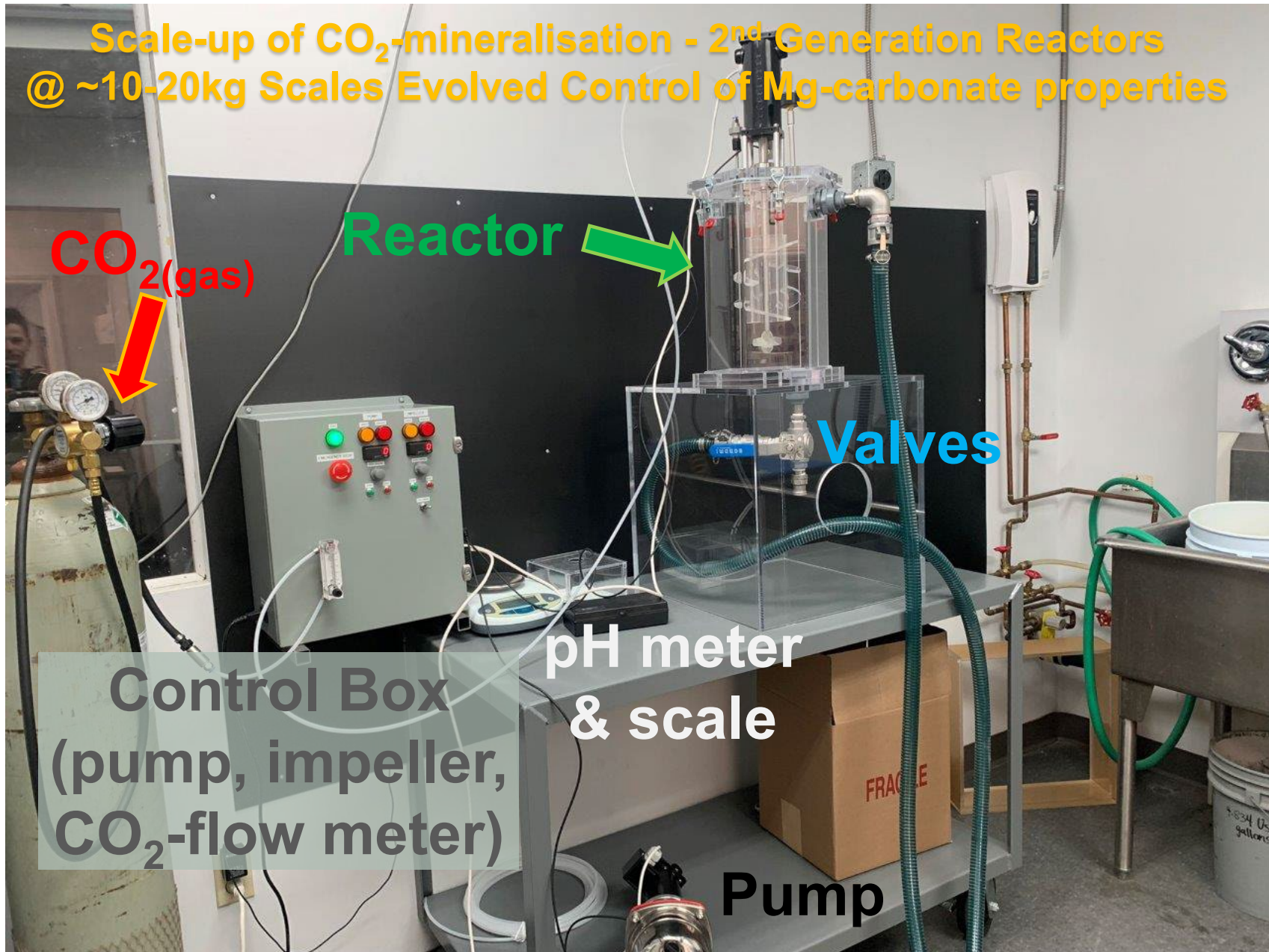
WP4: Determining viscosity in CO₂-mineralisation solutions with quasi-elastic neutrons



Key Points:

- Quasi-elastic scattering provides real-time tracking of viscosity. Changes ongoing Differing 'elasticity' values for the solutions are related to differing bulk viscosity and aggregation present and evolving...
- In this case, precise resolution of H₂O-solvent motions and differing aggregations of solutes.

Scale-up of CO₂-mineralisation - 2nd Generation Reactors @ ~10-20kg Scales Evolved Control of Mg-carbonate properties



CO₂(gas)

Reactor

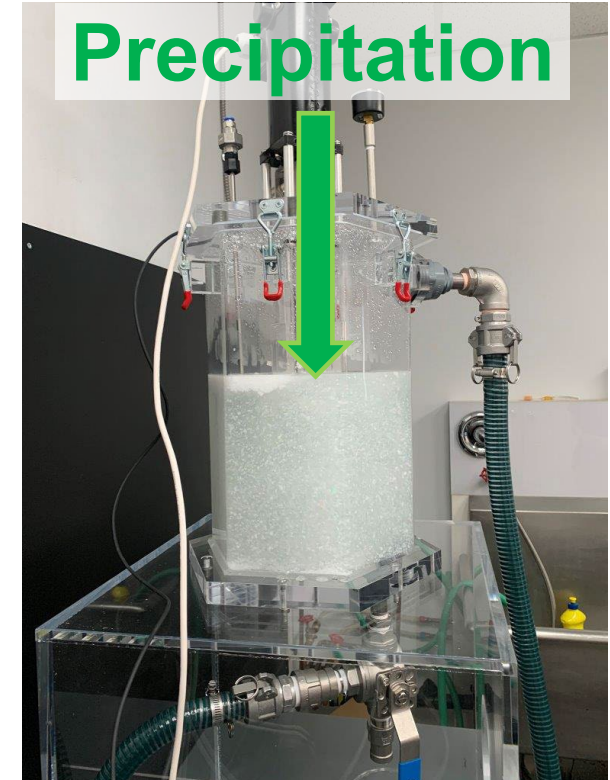
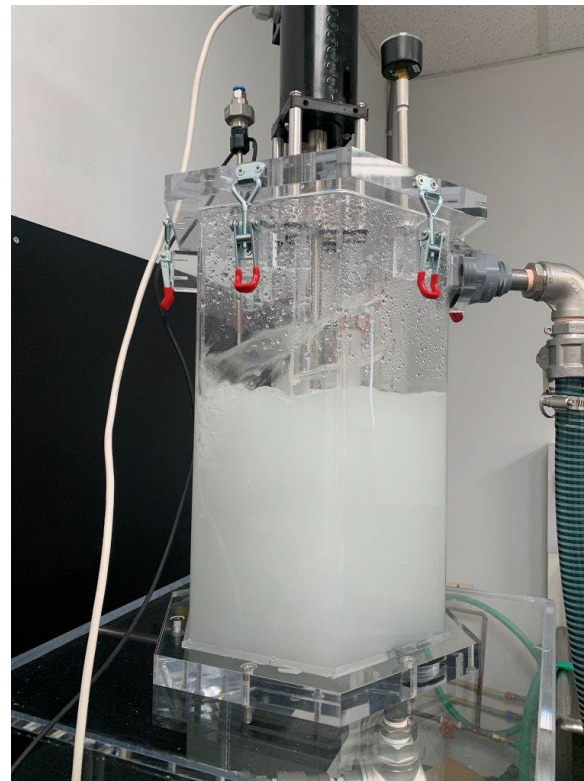
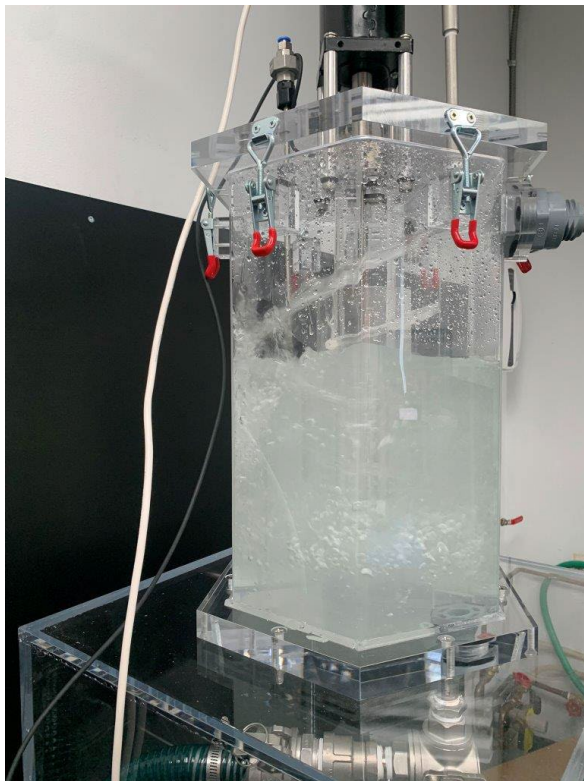
Valves

Control Box
(pump, impeller,
CO₂-flow meter)

pH meter
& scale

Pump

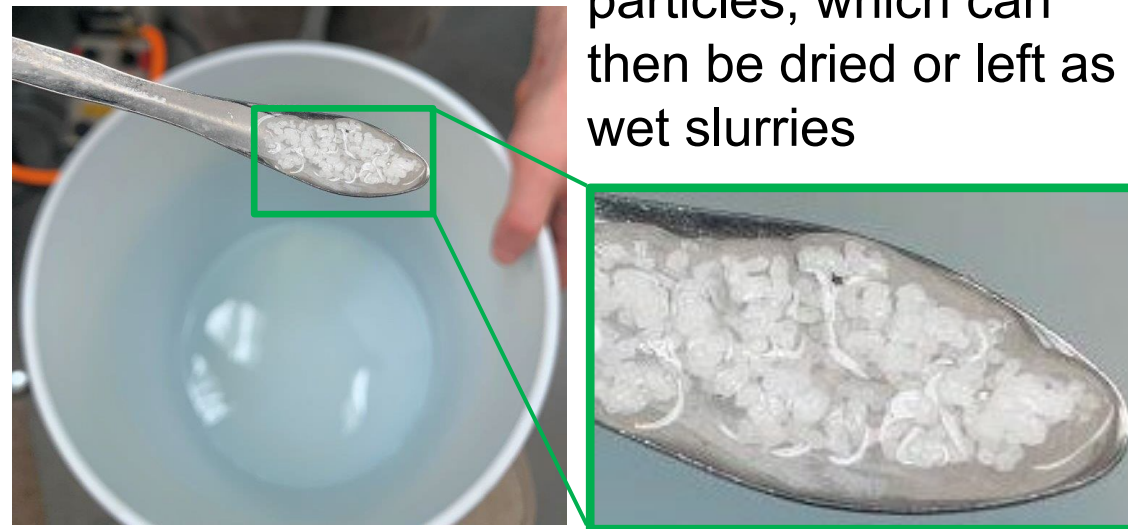
WP4: Scale-up: Here to 10+kg / hour. Visualising Bulk Mineralisation



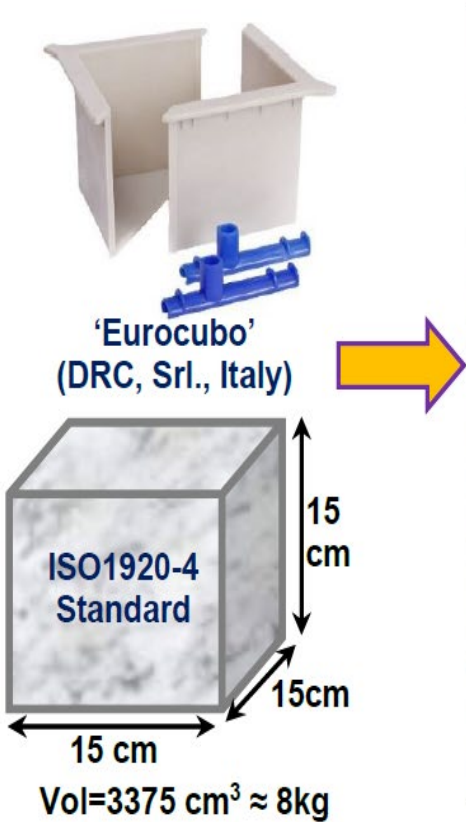
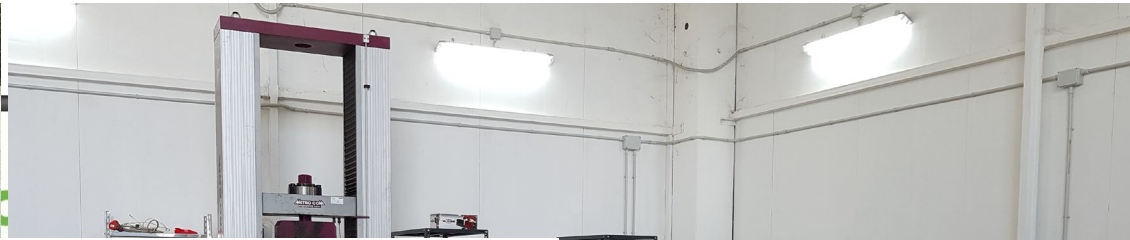
WP4: Scale-up: Precipitates form aggregates & clusters



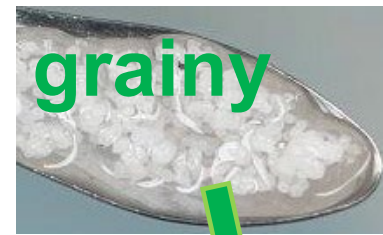
- Precipitates initially form flocculates
- Aggregates and clusters begin to form then supra- aggregate structures: stalactite like cones
- Filtration generates fine carbonate particles, which can then be dried or left as wet slurries



WP4: Scale-up: What to do with carbonates? Infrastructure



WP4: Scaling-up the Scale-Up: Compatibility with Concrete

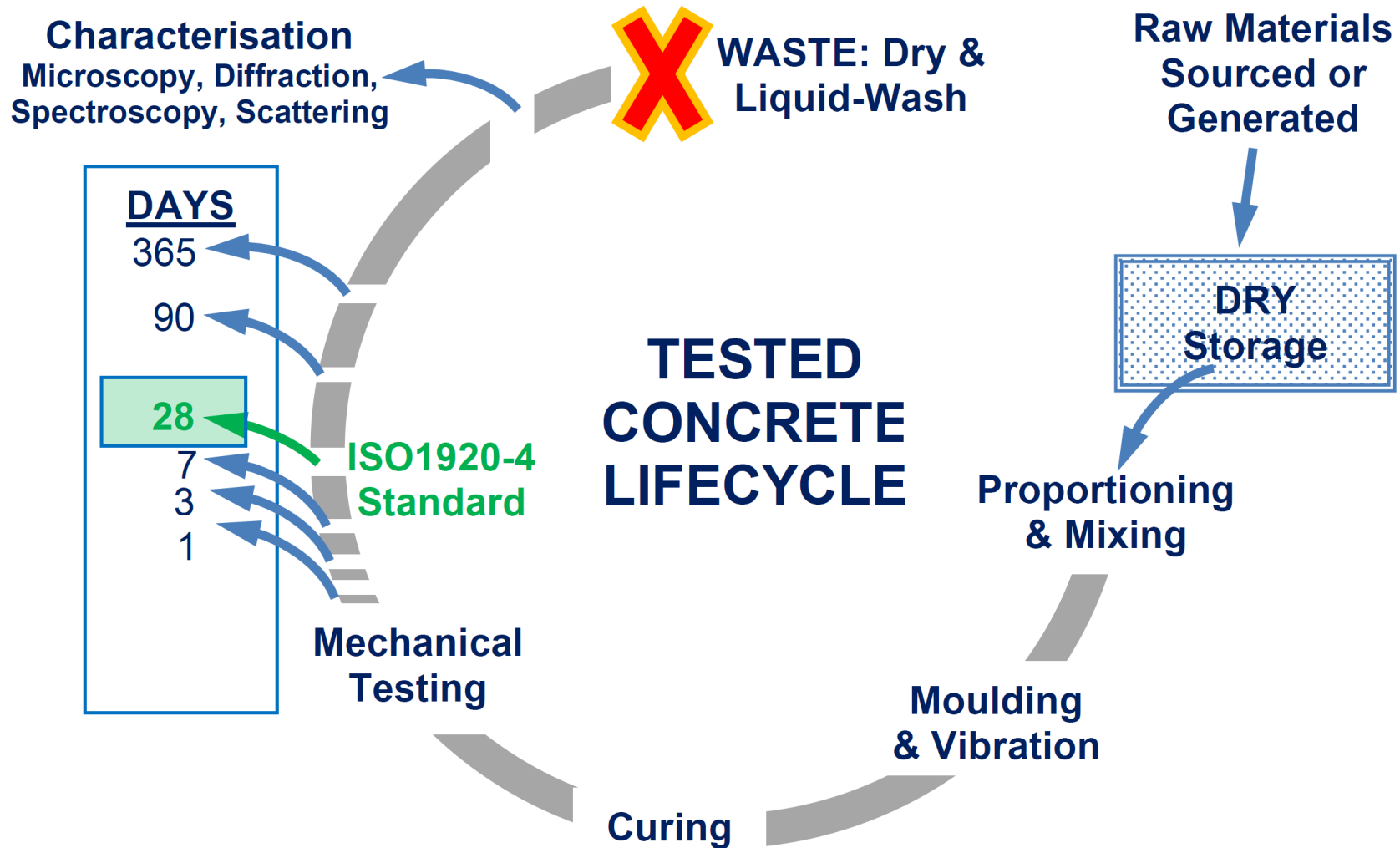


direct?

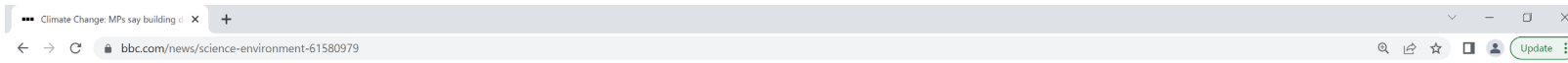


- Mg-CO_3 (s) can be directly added to concrete; replacing cement (up to 30%) = € £ \$ and lowers CO_2 footprint
- MgCO_3 s 'fixed' carbon further reduces CO_2 budget
- We have also added agricultural wastes = preventing their degradation and release of CO_2 , methane, etc.. (agriculture contributes ~ 9.5 gT/yr = $\sim 1/3$ of all emissions)

Sustainable Project Lifecycle



Demolition Waste – Same Fate as Fly Ash?



Climate Change: MPs say building demolitions must be reduced

By Roger Harrabin

BBC energy and environment analyst

🕒 26 May | 💬 Comments



<https://www.bbc.com/news/science-environment-61580979>

26 May, 2022

Acknowledgements



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